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dilution = moles of acid after dilution,  
and moles of acid =  $M \times V$  then,

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Chemistry If8766 Instructional Fair Inc - WordPress.com Molarity is expressed in units of moles per liter (mol/L). It's such

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a common unit, it has its own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

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where the subscripts "1" and "2" refer to the solution before and after the dilution, respectively. Since the dilution process does not change the amount of solute in the solution,  $n_1 = n_2$ . Thus, these two equations may be set equal to one another:  $[M_1L_1 = M_2L_2]$  This relation is commonly referred to as the

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dilution equation.

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Molarity is expressed in units of moles per liter (mol/L). It's such a common unit, it has its own symbol, which is a capital letter M. A solution that has the

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concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

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