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Observe how reactants and products interact in reversible reactions. The initial amount of each substance can be manipulated, as well as the pressure on the chamber. The amounts, concentrations, and partial pressures of each reactant and product can be tracked over time as the reaction proceeds toward equilibrium.

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In the equilibrium and concentration gizmo, you learned that you can predict the direction of a reaction by comparing the reaction quotient ( $q_c$ ) with the known equilibrium constant  $K_c$ . you can do the same thing using partial pressures: 1

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the pressure on the chamber. The amounts, concentrations, and partial pressures of each reactant and product can be tracked over time as the reaction proceeds toward equilibrium.

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## Answer Key -

## Maharashtra

Equilibrium occurs when two opposing processes occur at the same rate, leading to no net change. In the Equilibrium and Concentration Gizmo™, you will investigate how equilibrium can occur in chemical reactions. To begin, check that Reaction 1 is selected. Set Moles NO<sub>2</sub> to 8 and Moles N<sub>2</sub>O<sub>4</sub> to 0.

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given reaction at a

constant temperature,

no matter the starting

concentrations. That is

why  $K$  is known as the

equilibrium constant.

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PDF equilibrium

Equilibrium occurs when two opposing processes occur at the same rate, leading to no net change. In the Equilibrium and Concentration Gizmo™, you will investigate how equilibrium can occur in chemical reactions. To begin, check that Student Exploration Sheet:

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